

INTERNAL ENERGY:
 "SUM OF KE + PE OF ALL PARTICLES IN A SUBSTANCE."

- RAISE TEMP → KE
- CHANGE STATE → PE

CHANGING TEMP: (AB)

$$E = m \times SHC \times \Delta T$$

SPECIFIC HEAT CAP (C)

* IF SOLID MELTS AND TEMP ↑

$$E = m ((SHC \times \Delta T) + SLH)$$

* IF 2 SUBSTANCES TOUCH, REACH COMMON TEMP e.g.

$$m_1 c_1 (T - T_1) = m_2 c_2 (T - T_2)$$

* CONDUCTION: HEAT TRANSFERRED THROUGH VIBRATIONS.

* CONVECTION: HOT SUBSTANCE RISES, COLD SUBSTANCE FALLS

* RADIATION: INFRA-RED EM WAVE ABSORBED.

$$pV = \frac{1}{3} N m c_{rms}^2 = NkT$$

$$E_k = \frac{1}{2} m c_{rms}^2 \therefore E_k = \frac{3}{2} kT$$

BROWNIAN MOTION: RANDOM MOTION - PROVED BY SMOKE PUSHED AROUND BY AIR.

CHANGING STATE:

$$E = m \times SLH$$

SPECIFIC LATENT HEAT (L)
 (FUSION/VAPOURISATION)

Kinetic Theory:

Assumptions

- R - Random motion
- A - Attraction (none)
- V - Volume (negligible)
- E - Elastic collisions
- D - Duration of collisions short



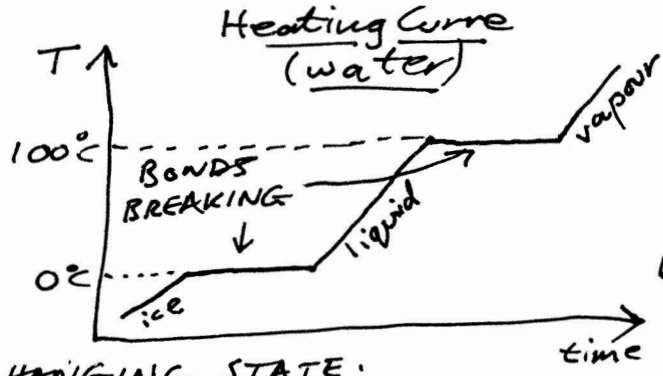
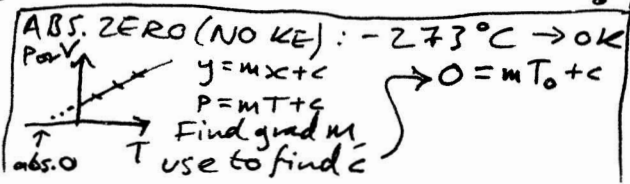
$$pV = \frac{1}{3} N m c_{rms}^2$$

(c_{rms}^2 = mean square speed, $m^2 s^{-2}$)

$$p \times \frac{1}{3} \left(\frac{Nm}{V} \right) c^2 = \frac{1}{3} p c^2$$

HEAT PUMP COP = $\frac{T_H}{T_H - T_C}$

FRIDGE COP = $\frac{T_C}{T_H - T_C}$



WHEN CHANGING STATE, ENERGY/HEAT DOES NOT INCREASE KE, ONLY PE BY BREAKING PHYSICAL BONDS (NOT CHEMICAL).

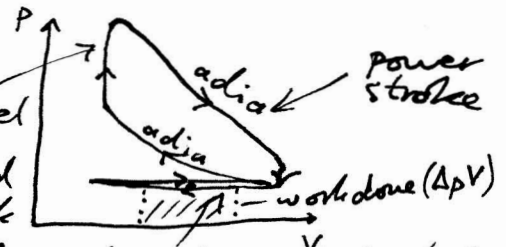
Isothermal: T constant
 $\therefore p_1 V_1 = p_2 V_2 \quad \Delta U = 0$

Isobaric: P constant $\frac{V_1}{T_1} = \frac{V_2}{T_2}$

Isochoric: V constant $\frac{p_1}{T_1} = \frac{p_2}{T_2}$

Adiabatic: $Q = 0 \therefore \Delta U = -W \quad p_1 V_1^\gamma = p_2 V_2^\gamma$

Otto Cycle: Petrol



HEAT ENGINE $\epsilon = \frac{T_H - T_C}{T_H}$

* THERMAL $\epsilon = \frac{\text{INDICATED P}}{\text{INPUT P (fuel)}}$

* MECH $\epsilon = \frac{\text{BRAKE (OUTPUT) P}}{\text{INDICATED P}}$

* OVERALL $\epsilon = \frac{\text{BRAKE P}}{\text{INPUT P}}$

GASES

Boyle's: $P \propto \frac{1}{V}$

Charles's: $V \propto T$

Pressure: $P \propto T$

Gas Law: $pV = NkT$

$pV = nRT$

1st Law of thermo:

$Q = \Delta U + W$

↑ ↑ ↑
 heat into E work done (expansion)

2nd law:

Heat cannot be converted into work unless it flows from a hot space to a cold one (aka no engine is 100% efficient)

x # cylinders
 x cycles per sec
 = INDICATED P.